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share a pair of electrons.
molecule. neutral group of
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together by one or more
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valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.6
 Chemical Bonding
 chemical bond in which two atoms share a pair of valence electrons. Define polyatomic ion: A covalently bonded group of atoms that has a positive or negative charge and acts as a unit. Name _____

_____ Chapter
 6 Notes - Chemical
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A mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together 6-1
 Introduction to Chemical Bonding I. Types of Chemical Bonding A. Ionic Bonding 1. Chapter 6 Review Chemical Bonding Answers Pdf a chemical bond resulting from the sharing of an electron pair between two atoms. nonpolar-covalent bond. a covalent bond in which the bonding electrons are shared equally by the bonded atoms, resulting

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ionic. type of bond that holds cations and anions together. covalent. bond in which two atoms share a pair of electrons.

molecule. neutral group of atoms that are joined together by one or more covalent bonds. polar.

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CHAPTER 6 REVIEW

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SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

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A chemical bond in which two atoms share a pair of valence electrons. Define polyatomic ion: A covalently bonded group of atoms that has a positive or negative charge and acts as a unit.

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6.2 Covalent Bonding. The attractions between the shared electrons and the protons in each nucleus

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